

Trace Element Assessment in Fingernails of Adult Females

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ABSTRACT

Trace elements play a significant role in giving nutritional benefits to the body because they act as essential cofactors for all physiological processes. However, there are some trace elements which may bring more harm than good when entering the human body. Due to its ability to incorporate trace elements in an amount that is proportional to an individual's dietary intake and environmental exposure, human fingernails are suitable biomarkers in assessing the health status of an individual as they reflect on the trace element concentration present in the body. This study has analysed fingernail samples of 23 adult females residing in Kuching and Kota Samarahan, Sarawak, Malaysia for four elements, namely Cd, Cu, Pb and Zn. By using flame atomic absorption spectroscopy (FAAS), the mean elemental concentrations found in fingernail samples of research participants were 171.8 ± 33.8 $\mu\text{g/g}$ for Zn, 27.8 ± 14.8 $\mu\text{g/g}$ for Cu and 2.64 ± 0.94 $\mu\text{g/g}$ for Pb. Cd concentrations were not able to be detected as they were below the detection limits. A standard reference material, NIST 1568b Rice Flour was used to verify the methods used in elemental analysis using FAAS. Independent t-test which was used to compare the means of Zn and Cu between vegetarians and non-vegetarians showed no significant differences for both elements. Moreover, correlation analysis showed negative correlations between Cu/Zn pair and Pb/Zn pair, whereas significant positive correlation was obtained for Cu/Pb pair. The overall data from this study showed good agreements with data obtained from studies in other countries. Therefore, the current data in this study represents the latest background elemental concentrations in fingernails of the residents in Kuching and Kota Samarahan, Sarawak.

Keywords: Adult females, fingernails, flame atomic absorption spectroscopy (FAAS), trace elements

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INTRODUCTION

Trace elements are known as dietary minerals which are needed in small quantities for ensuring normal physiological functions in the body of living organisms (Strachan, 2010). Trace elements are those which constitutes less than 0.01% of body mass. Being structural constituents of enzymes or cofactors, they play vital roles in the prevention of nutritional deficiencies, antioxidant defence, immune functions, regulation of gene expression and in the prevention of chronic diseases. Although most trace elements present in human bodies are essential elements produced naturally by the body itself, it is also possible for these elements to be taken in from the external environment. This leads to the increase in trace element concentrations in the body. Other than that, the uptake of environmental elements, such as heavy metals, into the human body is possible by three main routes, namely dermal absorption, inhalation and ingestion. Thus, upon high exposure to possible sources of environmental elements and heavy metals, the human body may contain high concentrations of trace elements which can be toxic and lead to negative health effects to the exposed individual (Esteban & Castano, 2008; Parizanganeh, Zamani, Bijnavand & Taghilou, 2014). Human biomonitoring (HBM) is defined as a scientific technique that allows researchers to investigate how much of environmental substances have entered the human body, besides looking at how these exposures vary over a certain period of time. The biological samples, such as blood, breastmilk, saliva, urine and nails, are collected, analysed and are then compared with suitable reference values in order to assess exposure and health risk of the exposed subjects (Angerer, Aylward, Hays, Heinzow & Wilhelm, 2011; Esteban & Castano, 2008; Waseem & Arshad, 2016). In this study, fingernail samples were chosen mainly due to its easy sample collection and storage, besides having the advantage of reflecting elemental concentrations in the body from a period of 12 to 28 months (Li *et al.*, 2012).

To the best of current knowledge, there has yet to be any published research on the proposed study area, namely Kuching and Kota Samarahan, Sarawak, Malaysia. As mentioned in Li *et al.* (2012), the scientific data for elemental nail concentrations are still limited even with the findings obtained from previous studies throughout

the years. It was further reported that most of the previous researches relied on data findings from other studies which were scattered worldwide. Therefore, the objective of this study is to determine the concentrations of cadmium (Cd), copper (Cu), lead (Pb) and zinc (Zn) in fingernails of adult females. These results were also compared with literature values from other countries.

MATERIALS & METHODS

Sample Collection and Washing

A total of 23 sets of fingernail samples were collected from healthy adult females who reside in Kuching or Kota Samarahan, Sarawak areas. These research participants were in the age range of 18 to 60 years old, in which the majority were university students. Among these participants, three of them were vegetarians whereas none were smokers. Fingernail samples (from 10 fingers) were collected from the participants using clean stainless-steel clippers and were stored in polyethylene bags. A minimum of three samples were collected from each participant, commonly at an interval of two to three weeks. Each individual was given a survey form, which was used to obtain required information including age, occupation, area of stay and her dietary habits, specifically whether she was a vegetarian or non-vegetarian and smoker or non-smoker.

Each batch of fingernail sample was washed using the methods reported in Wee and Ebihara (2017). Upon transferring samples into a beaker, acetone was added just enough to cover the nails and left aside for 20 minutes. This step was then repeated with deionized water for 20 minutes, followed by another washing with acetone for another 20 minutes. After washing, fingernail samples were transferred into a crucible before being placed in the drying oven (66 °C) for 30 minutes. The samples were fully dried when nails have achieved constant weight (Nyambura, 2012).

Acid Digestion and Analysis

Prior to acid digestion of fingernail samples, the weight of each set of nail sample was determined first. The acid digestion of samples was done using nitric acid, HNO₃ in a fume chamber. A hot plate was used as the heating platform. Next, 5 mL of 65% HNO₃ was added into each beaker containing nail samples, before heating the beakers at medium temperature. As nail samples were digested, light to dark yellow solution were formed. The sample solutions were left to digest and evaporate until about less than 1 mL of solution was left. Each set of samples averagely took 1 hr to 1 hr 45 minutes for digestion. When the sample had cooled down, they were diluted with 20 mL of deionized water. They were then filtered directly into respective sample bottles using 0.01 mm filter papers. When this step was done, the sample bottles were sealed tightly and kept in the refrigerator at 4 °C until FAAS analysis. These samples were kept not longer than 48 hours before analysis.

Analysis of Standard Reference Material, NIST 1568b Rice Flour

In order to ascertain the accuracy of method used in this experiment, the analysis of NIST 1568b Rice Flour was performed using the FAAS. A 0.5 g of the NIST 1568b Rice Flour were used, with the preparation methods similar to that of fingernail samples.

Statistical Calculation and Analysis

The Limit of Detection (LOD) was calculated using the formula $LOD = X_{bl} + 3S_{bl}$ where X_{bl} is the mean concentration of sample blanks ($n = 3$) and S_{bl} is the standard deviation of the sample blanks (Shrivastava & Gupta, 2011). Moreover, the results obtained from this study were analysed using IBM SPSS Statistics version 23.0. Firstly, the independent t-test was used to compare the means of each element between vegetarian and non-vegetarian participants. Secondly, correlation analysis was done to observe the relationship between the elements assessed in this study.

RESULTS & DISCUSSION

Analytical Quality Control

Table 1 shows the mean concentrations and standard deviations of blank samples used, including the experimental LOD.

Table 1. The experimental detection limits of FAAS in this study.

Element	λ (nm)	Mean concentration of blank samples ($\mu\text{g/g}$)	Limit of Detection, LOD ($\mu\text{g/g}$)
Zn	213.9	0.419 \pm 0.245	1.15
Cu	324.8	0.413 \pm 0.272	1.23
Pb	217.0	0.016 \pm 0.059	0.19
Cd	228.8	0.079 \pm 0.036	0.19

The results for Cd, Cu, Pb and Zn concentrations in NIST 1568b Rice Flour are as shown in Table 2, alongside the NIST certified values. These elemental concentrations were expressed as mean \pm standard deviation of three independent measurements.

Table 2. Comparison of elemental concentrations in this work with NIST 1568b certified values.

Element	NIST 1568b Rice Flour (in $\mu\text{g/g}$)		p- value
	Experimental Concentration	Certified Concentration	
Zn	21.69 \pm 1.32	19.42 \pm 0.26	0.25
Cu	2.39 \pm 0.04	2.35 \pm 0.16	0.40
Pb	^{a)} -	0.008 \pm 0.003	-
Cd	^{b)} -	0.0224 \pm 0.0013	-

^{a) & b)} Value less than instrument experimental detection limit.

The values observed between certified and experimental concentrations in this work showed good agreements for Cu and Zn, however the concentrations of Cd and Pb were below the experimental detection limit of FAAS. The values for Cu and Zn did not show any significant differences from certified concentrations as both p-values, which were obtained from one-sample t-test, were more than 0.05. In a report by Gholami, Behkami, Zain and Bakirdere (2016), it was stated that the values obtained for Cd, Cu, and Zn were $0.0219 \pm 0.0012 \mu\text{g/g}$, $2.34 \pm 0.09 \mu\text{g/g}$ and $18.72 \pm 0.51 \mu\text{g/g}$ respectively for NIST 1568b Rice Flour. The values obtained in this study were in good agreement with the certified values and those reported in Gholami *et al.* (2016). Thus, results obtained are acceptable for further discussions.

Elemental Concentrations in Fingernails

In this study, Cu and Zn could be detected in all fingernail samples, Pb only in seven samples whereas Cd concentrations were below the detection limits for all samples. Among the three elements detected, it was observed that Zn was the most abundant element with mean value of $171.8 \pm 33.8 \mu\text{g/g}$, followed by Cu with mean concentration of $27.8 \pm 14.8 \mu\text{g/g}$. Meanwhile, Pb had a mean value of $2.64 \pm 0.94 \mu\text{g/g}$. The overall elemental concentration of all samples are shown in Table 3. The standard deviation values used in these results were based on the triplicates of fingernail samples obtained, except for the case of P22 and P23 in which their standard deviations were based on two sets of fingernail samples.

The results of Zn concentrations in this study were in accordance to previous findings from literatures. As reported in Chaudhary, Ehmann, Rengan and Markesbery (1995), the Zn concentrations were found to have a mean value of $147.0 \mu\text{g/g}$, which was lower than that of this study. The Zn concentrations for three vegetarian participants were $145.6 \pm 27 \mu\text{g/g}$, $132.8 \pm 26.8 \mu\text{g/g}$ and $207.7 \pm 51.7 \mu\text{g/g}$. Other than that, the mean Zn concentrations of adult females in Kuching and Kota Samarahan were found to be lower than that of adult females in North, East and West Sudan, in which the Zn concentrations were reported to be $638 \pm 172 \mu\text{g/g}$, $667 \pm 127 \mu\text{g/g}$ and $885 \pm 127 \mu\text{g/g}$ respectively. As further explained in Ebrahim *et al.* (2011), one of the factors contributing to the high concentrations in Sudan was the dietary habits of people in these areas. Comparing the results of Zn levels in this study with that in other countries, the minimum range for Zn was not as low as the results reported. For instance, Rodushkin and Axelsson (2000) indicated in their study on individuals in Sweden that the minimum Zn concentration was $80.0 \mu\text{g/g}$, while Samantha, Sharma, Roychowdhury and Chakraborti (2004) reported minimum Zn concentrations of $72.8 \mu\text{g/g}$ in India and $48.2 \mu\text{g/g}$ in Nigeria (Olabanji *et al.*, 2005).

King, Shames and Woodhouse (2000) mentioned that zinc concentrations in normal human body occurs in the range of $107 \mu\text{g/g}$ to $231 \mu\text{g/g}$. If zinc concentrations are constantly below $22.0 \mu\text{g/g}$ for a long period of time, that particular individual may be having zinc deficiency, which leads to various diseases. However, a report by Kanabrocki, Kanabrocki, Greco, Kaplan and Oester (1979) has shown that the Zn concentrations had a range of $12.0 \mu\text{g/g}$ to $184 \mu\text{g/g}$ (mean value of $88 \mu\text{g/g}$) for individuals who were affected by beriberi. In contrast, Zn concentrations in this work were reported to be apparently higher than the normal range of $93.0 \mu\text{g/g}$ to $129.0 \mu\text{g/g}$ as stated in Mahler, Scott, Walsh and Haynie (1970). In Mahler *et al.* (1970), elevated concentration of Zn levels was found in patients who were affected by diseases such as uraemia, which had a mean concentration of $478.0 \mu\text{g/g}$, and patients who were undergoing the dialysis treatment, which had a mean concentration of $445.0 \mu\text{g/g}$. Therefore, it can be concluded that the levels of Zn in fingernail samples of all participants in this study were within the recommended range, besides not being subjected to any Zn deficiency or toxicity.

The Cu concentrations of the three vegetarian participants in this study were $18.1 \pm 0.51 \mu\text{g/g}$, $23.1 \pm 3.16 \mu\text{g/g}$ and $38.8 \pm 4.57 \mu\text{g/g}$ respectively. These values were comparably higher than that of Cu concentrations obtained among vegetarians of similar age group which were reported in Mehra and Juneja (2005), which observed values

of $8.96 \pm 1.35 \mu\text{g/g}$. Based on literature values of Cu from other countries, most of the concentrations found present in fingernail samples were less than $10 \mu\text{g/g}$. For example, Nowak and Chmielnicka (2000) reported that Polish volunteers had mean Cu concentrations of $7.0 \mu\text{g/g}$ in their fingernails, Canada volunteers gave mean values of $9.2 \mu\text{g/g}$ (Li *et al.*, 2012), Saudi Arabia adults had values of $9.01 \pm 1.03 \mu\text{g/g}$ (Momen, Khalid, Elsheikh & Ali, 2015) and the participants in Japan gave results of $5.18 \pm 1.72 \mu\text{g/g}$ (Wee & Ebihara, 2017). When the results of this work were compared with the study done in Malaysia previously, Cu levels in nails of volunteers in Kuching and Kota Samarahan were very high. Ghazali *et al.* (2013), which researched on trace element concentrations in the nails of elderly people living in FELDA settlement in Selangor, found out that Cu concentrations were $4.81 \pm 1.16 \mu\text{g/g}$.

Table 3. Elemental concentrations (in $\mu\text{g/g}$ dry mass) in fingernails of adult females. Vegetarian participants were marked with asterisk (*).

Participants	Concentrations (in $\mu\text{g/g}$ dry mass)			
	Zn	Cu	Pb	Cd
P1	161.8 ± 12.5	49.09 ± 1.96	< 0.19	< 0.19
P2	220.2 ± 43.2	39.41 ± 5.52	< 0.19	< 0.19
P3	196.9 ± 1.8	22.18 ± 3.14	< 0.19	< 0.19
P4	202.5 ± 69.3	39.23 ± 4.86	< 0.19	< 0.19
P5	159.6 ± 39.4	19.73 ± 4.89	< 0.19	< 0.19
P6	169.8 ± 25.7	15.30 ± 1.55	< 0.19	< 0.19
P7	170.1 ± 22.4	24.50 ± 3.27	2.48 ± 0.04	< 0.19
P8	197.1 ± 9.5	8.37 ± 0.55	< 0.19	< 0.19
P9	239.7 ± 46.6	10.04 ± 1.56	< 0.19	< 0.19
P10	152.9 ± 7.5	6.99 ± 1.11	< 0.19	< 0.19
P11	167.0 ± 9.7	6.33 ± 1.91	< 0.19	< 0.19
P12	143.2 ± 19.4	7.92 ± 1.43	< 0.19	< 0.19
P13*	145.6 ± 27.0	38.78 ± 4.57	< 0.19	< 0.19
P14	206.9 ± 36.4	36.67 ± 2.02	< 0.19	< 0.19
P15*	132.8 ± 26.8	18.09 ± 0.51	2.51 ± 0.04	< 0.19
P16	134.5 ± 17.8	31.29 ± 1.75	4.04 ± 0.41	< 0.19
P17	115.7 ± 8.9	55.39 ± 1.96	3.66 ± 0.60	< 0.19
P18	185.5 ± 7.3	35.84 ± 3.22	2.10 ± 0.03	< 0.19
P19	197.2 ± 42.7	48.84 ± 2.01	< 0.19	< 0.19
P20	123.9 ± 7.0	33.93 ± 1.48	1.25 ± 0.24	< 0.19
P21	129.6 ± 0.5	42.53 ± 6.14	< 0.19	< 0.19
P22*	207.7 ± 51.7	23.09 ± 3.16	< 0.19	< 0.19
P23	192.0 ± 4.5	25.70 ± 2.74	2.48 ± 0.25	< 0.19

However, there were other studies which have found high levels of Cu in the human body which also used fingernails as biomarkers. Ashraf, Jaffar and Mohammad (1994) have reported a mean Cu concentration of $22.4 \mu\text{g/g}$, whereas in the following year, another publication obtained a result of $20.8 \mu\text{g/g}$ (Ashraf, Jaffar, Anwer & Ehsan, 1995). This particular study was done among occupationally non-exposed Russian adults who lived in urban areas. The results of high Cu levels were also supported by Ebrahim *et al.* (2011), whereby healthy individuals of Sudan gave mean Cu concentrations of $73.8 \mu\text{g/g}$. According to Bost *et al.* (2015), there are still uncertainties in the medical field regarding the Cu reference values for humans. It is known that the daily intakes of Cu below 0.8 mg/day commonly leads to Cu deficiency, whereas the intakes of more than 2.4 mg/day would

lead to consistent net gains of this element in the body. Playing an important role in the synthesis of haemoglobin, carbohydrate metabolism and acting as a co-factor for many redox enzymes, Cu levels in the body is agreed to be highly dependent on the dietary intakes of each individual. For example, meat products such as liver, beef, and kidney are known to have high Cu levels in the range of 2.1 mg/kg to 4.3 mg/kg. This comes to show in cereal products, vegetables and seafood whereby the Cu concentrations present ranges in about 0.6 mg/kg to 16.6 mg/kg, 0.48 mg/kg to 16.0 mg/kg, and 0.3 mg/kg to 16.0 mg/kg respectively. Besides dietary habits, this element is also able to spread through the intake of birth control pills or nutritional deficiencies supplements, as well as spread environmentally, such as through hot water pipes, construction areas and dusts from mining sites (Ashish, Neeti & Himanshu, 2013).

Lead, Pb is a type of heavy metal which is commonly present in 8 mg/kg to 20 mg/kg in the earth's crust. It is not an essential element in the human body, therefore, when someone is highly exposed to this element for long periods of time, he or she may be said to be contaminated with lead (WHO, 2015). In this study, the LOD for the analysis of Pb was calculated as 0.19 µg/g. The results of Pb in this study had an overall mean value of 2.64 ± 0.94 µg/g ($n = 7$), in which all participants were non-vegetarians except for one. Moreover, the results obtained in this experiment were in accordance to the available Pb concentrations which were found in similar literatures as discussed below. In a fingernail analysis study by Kanabrocki *et al.* (1979), the results of Pb for healthy adult female ages between 23 years old to 60 years old were in the range of 1.7 µg/g to 24.0 µg/g. Moreover, other similar studies have had low Pb concentrations of 1.35 µg/g to 1.98 µg/g in non-smoking adults (Afridi *et al.*, 2008), 0.72 µg/g in healthy adult volunteers (Gouille *et al.*, 2009) and 1.046 µg/g in occupationally non-exposed Russian adults (Skalny *et al.*, 2015). These examples indicated that Pb concentration in this study was considered as low levels.

However, when the results of Pb in adult females in this study were compared to that of another Malaysian study, the results in the latter research was reported to be higher, which was 6.2 ± 5.2 µg/g (Ghazali *et al.*, 2012). This may be due to the occupational exposure of research participants involved in their study. Since they were taking nail samples from adults working in an agricultural farm, they might have been exposed to high levels of pesticides and fertilizers, which caused the higher Pb concentrations.

Although high Pb concentrations are constantly related to toxicity in the human body, a research by Mehra and Juneja (2005) had obtained Pb levels of 20.21 ± 6.17 µg/g in their control group. This control group were healthy individuals, in which the obtained Pb values were used to compare with the Pb levels obtained from nail samples of individuals with health disorders. Those with health disorders such as skin disease, hypotension, hypertension and diabetes actually shown variations of Pb concentration results. These results were either higher than or lower than the results of healthy individuals. In another report, it was indicated that high Pb concentrations in fingernail samples were only found in those who were exposed for long periods of time through their occupation. The results in Abdulrahman, Akan, Chellube and Waziri (2012) have proven that workers of a welding factory had very high Pb levels, 19.45 µg/g to 55.67 µg/g. Due to this, the workers in this particular research were said to be contaminated with Pb, in which this situation may lead to negative health effects. Through comparison with literature, Pb concentrations in fingernail samples show variations depending on the health status as well as the occupational-exposure of individuals. This can be seen through the concentrations of Pb among adult females in Kuching and Kota Samarahan which were still within the acceptable range. The concentrations showed that they were in good agreements to literature values of healthy and non-smoking participants (Afridi *et al.*, 2008; Gouille *et al.*, 2009), besides having the results in this study not as high as Pb concentrations found in occupationally-exposed individuals (Abdulrahman *et al.*, 2012; Ghazali *et al.*, 2012; Skalny *et al.*, 2015).

Statistical Analysis

In this study, the independent t-test was done for comparing the means of Zn and Cu between vegetarians and non-vegetarian participants. Pb was not analysed as there were insufficient information for any comparison to be done. Since there were only three vegetarian participants who were in the age range of 20 – 29 years old, their results were only compared with the results of non-vegetarians in similar age group. The Cu and Zn concentrations in vegetarian and non-vegetarian participants were compared as both elements were reported to have significant differences between these two groups (Hunt, 2003; Yoshida, Ogi, & Iwashita, 2011). Yoshida *et al.* (2011) mentioned that Cu intake in vegetarian women were slightly higher than non-vegetarian women as Cu is present in higher content in vegetables. Additionally, Hunt (2003) explained that since Zn is present in higher content in meats, the absorption of Zn in vegetarians were considerably lower.

In this study, it was found that there were no significant differences between Zn and Cu concentrations present in vegetarian and non-vegetarian nail samples, with the p-value being 0.294 and 0.133 respectively. These results differed from those reported in Mehra and Juneja (2005) as in their research, there were significant differences between Zn concentrations of these two groups for both Zn and Cu.

According to Samantha *et al.* (2004), correlation coefficients are used to show potential relationships or any interactions between two elements which are assessed in fingernail samples. The correlation coefficients, in this case, the Pearson correlation coefficient, was used and represented as a correlation matrix which is shown in Table 4.

Table 4. Correlation matrix among elemental concentrations in fingernail samples. The number in bold indicates that the concentrations are significant at $p = 0.05$.

	Zn	Cu	Pb
Zn	1		
Cu	-0.155	1	
Pb	-0.358	0.445	1

A significant positive correlation was observed between Cu and Pb, in which $r = 0.445$. This showed that although Pb is a heavy metal which might become toxic to the body, there was a good balance between these two elements in the fingernail samples of adults in this study. Next, the pair of Zn with Cu, and Zn with Pb showed negative correlations of $r = -0.155$ and $r = -0.358$ respectively. Both pairs showed no significant differences at $p = 0.05$. The negative correlation between Zn and Cu was opposite to the results reported in Wee and Ebihara (2017), whereby their research has indicated that Zn and Cu had positive correlations. Unlike the explanations made in this literature, this study showed that Zn and Cu were not in good balance among the research participants. Instead, present results showed the presence of either one of these elements in high amounts might cause the deficiency of the other (Lech & Sadlik, 2007).

As for the pair of Zn and Pb, the results of this study were opposite to that in Mehra and Thakur (2016), in which the Zn/Pb pair gave strong positive correlation among their research volunteers who were healthy and acted as control group. However, based on Fadayon, Abdollahi, Nia and Ostovar (2013), the excess and deficiency of some essential trace elements like Zn, are more likely to be affected by the presence of heavy metals like Pb and Cd. It was further explained that heavy metals in excessive amounts in the body damages organs and disrupts the physiological homeostasis, indirectly disrupting trace element concentrations as well. Therefore, the results of this study are obviously different from studies done in other countries in terms of the correlation between Zn and Pb.

CONCLUSION

In this study, fingernail samples from 23 healthy adult females from Kuching and Kota Samarahan had been collected and analysed. The sample collection and preparation had been done accordingly, followed by the analysis of Cd, Cu, Pb and Zn using FAAS. Zn was found to be higher than other elements in all fingernail samples, with mean concentration of $171.8 \pm 33.8 \mu\text{g/g}$, followed by Cu with mean value of $27.8 \pm 14.8 \mu\text{g/g}$. Among the seven samples in which Pb was detected, the mean concentration was $2.64 \pm 0.94 \mu\text{g/g}$. Cd levels were too low to be detected by FAAS. Comparing Zn and Cu concentrations between vegetarian and non-vegetarian participants, no significant differences were found through the independent t-test. Nevertheless, significant correlation was observed for Cu/Pb pair whereas the analysis showed negative correlations for other pairs. The results obtained from this research could serve in providing baseline reference values for future studies for the people in Kuching and Kota Samarahan, Sarawak, Malaysia.

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