Operational Parameters Assessment of a Biomass-To-Fuel Gas Converter

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Abstract—This study presents the assessment on the existing mini-sized biomass-to-fuel gas (B2F) converter. Operational parameters, such as composition of after filtered residual liquid (AFRL), composition of raw combustible fuel gases produced using wood chips and coal as feedstock, and nearby incremental levels of carbon monoxide and suspended particulate during operation, are determined. Analysis results of the raw (unburnt) combustible fuel gases from low grade Balingian-Mukah coal and wood chips found that there were approximately 23-29% gasoline constituents; 15-30% diesel constituents; 1-16% CO₂; 10-13% O₂, CO, SO, NO and H₂O; 46-71% combustible gases. It was observed that the phenol contained in the AFRL recorded approximately 5,310 mg/L. Moreover, by-products, such as 1.5% tars and 13% AFRL, were produced. The result shows that a B2F converter is suitable to harvest the combustible gases that can be used as fuel for internal combustion engines. However, the amount of organics indicated that the AFRL should not be discharged directly into watercourses without pre-treatment.

Keywords: Gasification, Combustible gas, Coal, Wood chips

I. INTRODUCTION

In Malaysia's logging industries in 2009, wood waste alone gave a total production of logs, based on the total land area, of 18.27 million m³ [1]. From January to February 2009, Malaysia generated a free on board (FOB) value of more than RM 202 million through logging activities [2]. This generated a huge amount of wood waste that, potentially, could have given rise to sensitive environmental disposal issues. In 2013, the total current energy demand in Malaysia was 33.9 Mtoe; by 2020, the energy demand is expected to grow by 5.4% per annum to reach 83.5 Mtoe [3]. It is estimated that the oil reserves in Malaysia will last for 15 years and gas reserves for another 29 years [4]. Although efforts at renewable energy have been initiated by the Malaysian government, the response has not been positive [5]; regardless, renewable energy still has great growth potential.

It is well-known that gasification is a comparatively cheap method that can produce product gas with very low tar content [6]. According to Walawender's study of a model of gasifier (stratified and open top), the average composition of the gas on a dry basis is as thus: $H_2 = 15.1\%$; CO = 19.1%; $CH_4 = 2.5\%$; $CO_2 = 15.8\%$; small quantities of C_2H_6 , C_2H_2 and C_3H_4 and the remainder is N₂ [7]. Garcia-Bacaicoa built two stratified-closed top gasifiers: one for 50 kg/h and another for 200 kg/h of the same configuration. The average composition of the gas produced on a dry basis is in the following range of volume percentage: N₂ = 45-60\%; $CH_4 = 0.25-2.5\%$; $H_2 = 10-22\%$; CO = 13-25%; $CO_2 = 8-19\%$ and traces of C_2H_2 and C_2H_4 [8]. In addition, the burning of biomass will introduce chemicals, such as polycyclic aromatic compounds (PAC) and heavy metals [9].

To date, various types of biomass combustor are available, such as updraft, downdraft, cross-draft, fluidised-bed design features and others [10]. However, treatment technologies for gasification in terms of cost-effectiveness, local availability and serviceability of materials and equipment that require low technical skills for operation and maintenance are not available [11]. Therefore this study aims to investigate the feasibility of a B2F converter for fuel production.

A. B2F Converter

A B2F converter (Figure 1) at the 12th Mile Oya Road, Sibu (Malaysia) was developed, constructed and tested for drying and heating purposes in commercialised industrial scale applications, the mushroom culture industry, egg tray manufacturing facilities, powdered coconut milk production, and paper recycling facilities. For this system, solid feedstock (wood chips

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and Balingian-Mukah coal) are gasified in the updraft B2F converter, and the fuel gas or combustible gas is produced by air mixture to maximise the efficiency of the gasification process. The fuel gas is purified by filtration process before being used as low-medium heat value fuel for heating and drying purposes.



Figure 1: Simplified schematic diagram of B2F converter

B. Components of B2F Converter

A B2F consists of various components. The incinerator comprises of a layer of fire-proof brick as the inner wall and is surrounded by a layer of water compartment. A B2F converter is equipped with an incinerator, which includes an ash collection (Figure 2). A plurality of air inlets at the bottom and upper of incinerator are introduced to enhance pyrolysis and combustion conversion. In addition, the B2F converter is equipped with filters connected to the incinerator by a gas conveying pipe (Figure 3). The filters of the B2F converter are equipped with the following features (Figures 3): (1) an inlet connected with a gas conveying pipe, (2) an internal vertical duct at the middle of the filter, and (3) an outlet connection at the upper part of the internal vertical duct to direct the filtered gas for usage. The internal vertical duct of the filter is made up of a wider dimensional opening at the bottom end and forms a conical shape duct, as it moves further up before the vertical duct (Figure 3). There is an external chamber with a tap for removal of overflowed liquid in the filters (Figure 4). Finally, the incinerator includes an upper gas storage case for storing combustible gas before sending it to the gas conveying pipe.



Figure 2: Cross-sectional illustration of a B2F converter



Figure 3: Cross-sectional view of a filter unit



Figure 4: Cross-sectional view of the top part of a B2F converter

II. MATERIAL AND METHODS

A. Condensate AFRL measurement

Samples of the condensate AFRL, collected from the bottom of the filters, were analysed with a gas chromatography mass spectrometer (GCMS) (Model: Hewlett Packard HP 5890 Series II Plus). The analysed chemical composition of the samples (wood and coal) included the following.

- Inorganic analysis—Acid digest according to APHA 3030E and heavy metals (Sb, As, Ba, Be, Cd, Cr, Co, Cu, Pb, Mo, Ni, Se, Ag, V, Zn, B and Fe) analysis according to SW 846 610B.
- Mercury analysis were performed in accordance to APHA 3112B.
- Organic compounds analysis.

In each analysis, a 100 ml sample was extracted with 300 ml of hexane; the aqueous layer was discarded, and the collected hexane was evaporated to 1 ml by using a rotary evaporator; then, a 1.0 micro-litre of sample was injected into the GCMS with a temperature programme from 35° C - 250° C at 10° C/min (Column specification: 30m, 0.25mm, 0.25um). There are two analysis assumptions in this experiment; firstly, density of hexane extract is considered to be 1.0 g/ml; secondly, all phenol derivatives were detected as phenol in the standard method for phenol determination according to APHA 5530D.

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B. Exhaust gas measurement

In this experimental fieldwork, wood chips and Balingian-Mukah coal were used as feedstock. The amount of wood chips and Balingian-Mukah coal used in this experiment are approximately 130 kg, respectively. During operation, samples for combustible fuel gas were collected at the exhaust (tail pipe effluent) point of the B2F and at nearby areas (<20 meters radius from emission source) for chemical compounds, including CO, CH_4 , volatile organic compounds and suspended particulate matters. A sample was extracted with 2 ml of hexane; the aqueous layer was discarded, and the collected hexane was evaporated to 1 ml by using a rotary evaporator; then, a 1 micro-litre of sample was injected into GCMS with a temperature programme from $45^{\circ}C-250^{\circ}C$ at $5^{\circ}C/min$ (Column specification: 30m, 0.25mm, 0.25um).

III. RESULT AND DISCUSSION

A. Characteristics of Condensate AFRL

Table 1 shows that some of the inorganic and heavy metals contained in the condensate (AFRL) were not in compliance with Malaysia's sewage and industrial effluent discharge standards (Environmental Quality Act, 1974), with the inclusion of cadmium (0.08mg/L), nickel (1.0 mg/L) and lead (1.73mg/L). The finding in this work is in agreement with [12], which used vineyard residues as a fuel for domestic boilers. Ireland's results showed that a mixture of coal and sewage sludge combustion produced a higher concentration than a wood chip sample in terms of As, Cd, Cr, Cu, Ni, Pb, Se and Zn [9]. The level of phenol contained in the AFRL measured as high as 5,310 mg/L, which exceeded the 0.001 mg/L limit and did not comply with the standard. Thus, one of the reasons for this result is due, mainly, to the sample's contamination [13].

Parameter Concentration Standard Standard Mixture of coal in AFRL B and sewage Α sludge [9] (mg/L)(mg/L)(mg/L)Silver, Ag 0.005 N/A N/A Arsenic, As 19-20 < 0.007 0.05 0.10 Boron, B 119 1.0 4.0 Barium, B 0.29 N/A N/A Beryllium, 0.0008 N/A N/A Be Cadmium, Cd 0.08 0.01 0.02 2-3 Cobalt, Co 0.18 N/A N/A Chromium, 0.77 N/A N/A 21-32 Cr 0.20 53-120 Copper, Cu 0.04 1.0 Iron, Fe 0.375% 1.0 5.0 Molybdenum, < 0.001 N/A N/A Mo Mercury, Hg 0.001 0.005 0.05 Nickel, Ni 1.0 0.20 1.0 30-32 Lead, Pb 1.73 0.10 0.5 43-110 N/A N/A Antimony, Sb 0.11 Selenium, Se 0.06 N/A N/A 2 Thallium, TI 0.11 N/A N/A Vanadium, V < 0.001 N/A N/A Zinc, Zn 2.0 2.0 104-235 18 Phenol 0.001 1.0

Table 1: Inorganics and heavy metals in AFRL as compared to Standard A and Standard B (wood chips as feedstock)

Note * Standard A and Standard B refer to the parameter limits of effluent set in Environmental Quality (Sewage and Industrial Effluents) regulation, 1979 [14] *N/A means Not Available

B. Combustible Fuel Gases and Suspended Particulate Matters

Analysis results of the raw (unburnt) combustible fuel gases from low grade Balingian-Mukah coal and wood chips as feedstock are summarised in Table 2. It was found that there were approximately 23-29% gasoline constituents; 15-30% diesel constituents; 1-16% CO₂; 10-13% O₂, CO, SO, NO and H₂O; 46-71% combustible gases. This result shows that the B2F converter is suitable to harvest the combustible gases that can be used as fuel for internal combustion engines [15].

During operation, nearby (with a 20 meter radius) incremental pollutant levels measured 25-40 ppm CO, 6-10 ppm H₂S, 80-110 μ g/m³ total suspended particulates and <30 μ g/m³ PM10 and ash and charcoal (negligible amount); 10-13% H₂O

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condensate; 1-2% tars (Table 2). Results from Table 2 indicate the produced air pollution was under the optimum condition [16], [17] and [18].

Table 3 shows the primary composition of exhaust gas; specifically, wood chips as feedstock recorded approximately 3.5% O_2 , 16.1% CO_2 , >10,000 ppm CO, 271 ppm NO_x and 190 ppm SO. However, the key composition of exhaust gas from Balingian-Mukah coal as feedstock measured 9.3% O_2 , 10.1% CO_2 , >10,000 ppm CO, 65 ppm NO_x and 25 ppm SO. It was shown that coal as feedstock of a B2F converter produced relatively less pollutants in the exhaust gas, when compared to wood chips.

Table 2: Summary of unburnt combustible gas compositions and nearby incremental air pollutant generated by Balingian-Mukah coal and wood chips during operation

Fuel gases	Results						
Unburnt combustible gas composition	Approximately 23-29% gasoline constituents 15-30% diesel constituents 8-12% poly-aromatics 1-16% carbon dioxide (CO ₂) 10-13% of O ₂ , CO, SO, NO & H ₂ O						
	46-71% combustible gases (i.e. gasoline, diesel constituents and poly-aromatics)						
Nearby incremental air pollutant	25-40 ppm CO 6-10 ppm H ₂ S 80-110 μg/m ³ total suspended particulates and <30 μg/m ³ PM10 Negligible ash and charcoal 10-13% H ₂ O condensate 1-2% tars						

Table 3: Analysis of exhaust gas (combustion by-products of combustible fuel gas)

Fuel (Feedstock)	Oxygen %	CO2 %	CO ppm	NO ppm	NO ₂ ppm	NO _X ppm	SO ppm
Wood chips	3.5	16.1	>10,000	271	N.D	271	190
Coal	9.3	10.1	>10,000	65	N.D	65	25

*N.D means Not Detected

IV. CONCLUSIONS

This study presents the results of an operational parameters assessment on a B2F converter, from which the following conclusions can be drawn. Analysis results of the raw (unburnt) combustible fuel gases from low grade Balingian-Mukah coal and wood chips found that there were approximately 23-29% gasoline constituents; 15-30% diesel constituents; 1-16% CO_2 ; 10-13% O_2 , CO, SO, NO and H_2O ; 46-71% combustible gases. The level of phenol contained in the AFRL measured as high as 5,310 mg/L, which exceeded the 0.001 mg/L limit; therefore, it does not comply with the Standard A and Standard B established under the Environmental Quality (Sewage and Industrial Effluents) Regulations, 1979. Approximately 1.5% tars and 13% AFRL were produced as by-products. The result shows that the B2F converter is suitable to harvest the combustible gases that can be used as fuel for internal combustion engines. However, the amount of organics indicated that the AFRL should not be discharged directly into watercourses without pre-treatment.

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